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Chapter 11 Review Gases Section

CHAPTER 11 REVIEW Gases SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. b Pressure surf f a o c r e ce area. For a constant force, when the surface area is tripled the pressure is (a) doubled. (b) a third as much. (c) tripled. (d) unchanged. 2. d, c, a, b Rank the following pressures in increasing order. (a) 50 kPa (c) 76 torr (b) 2 atm (d) 100 N/m² 3.

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SECTION 1 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. b Pressure — orce For a constant force, when the surface area is tripled the surface area pressure is (a) doubled. as much. (c) ripld. 7-0 (d) unchanged. Rank the following pressures in increasing order. (c) 76 torr (a) 50 kPa O, OOİctbv-x

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Section 11.4 Dalton's Law of Partial Pressures Goals To describe the properties of mixtures of gases. To describe calculations that deal with mixtures of gases. In the real world, gases are usually mixtures. This section describes how mixing gases affects the properties of the resulting mixture.

Chapter 11 - Gases

462 Chapter 11 Gases Discovering the Relationships Between Properties If we want to explain why a weather balloon carrying instruments into the upper atmosphere expands as it rises, we need to consider changes in the properties of the gases (pressure, volume, temperature, or number of gas particles) inside and outside the balloon.

Chapter 11 Gases - An Introduction to Chemistry

Chapter 11 Review Gases Section CHAPTER 11 REVIEW Gases SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. b Pressure surf f a o c r e ce area. For a constant force, when the surface area is tripled the pressure is (a) doubled. (b) a third as much. (c) tripled. (d) unchanged. Page 6/9

Chapter 11 Review Gases Section 1 Answer Key

Start studying Chapter 11- Gases: Section 1: Gases and Pressure. Learn vocabulary, terms, and more with flashcards, games, and other study tools.

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Chapter 11- Gases: Section 1: Gases and Pressure ...

CHAPTER 11 REVIEW Gases SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances: increase a. temperature increases, volume stays the same decrease b. volume increases, temperature stays the same

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CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided 1 The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol (c) multiplied by 224 L (b) divided by the mass of 1 mol (d) divided by 224 L 2 For the expression $V = nR \frac{P}{T}$...

Chapter 11 Review Gases Section 1 Answers

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