

Chapter 9 Review Stoichiometry Section 1 Answers

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Chapter 9 Review Stoichiometry Section

CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N₂ are mixed with 12.0 mol of H₂ according to the following equation: $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$

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Stoichiometry. SECTION 1. SHORT ANSWER Answer the following questions in the space provided. 1. ____ The coefficients in a chemical equation represent the (a) masses in grams of all reactants and products. (b) relative number of moles of reactants and products. (c) number of atoms of each element in each compound in a reaction.

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Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. _____
The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N₂ are mixed with 12.0 mol of H

CHAPTER 9 REVIEW Stoichiometry

9-1 Introduction to Stoichiometry Composition Stoichiometry - deals with mass relationships of elements in compounds
Reaction Stoichiometry - Involves mass relationships between reactants and products in a chemical reaction I. Reaction Stoichiometry Problems A. Four problem Types, One Common Solution

Chapter 9 - Stoichiometry

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Stoichiometry. SECTION 2. PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. The following equation represents a laboratory preparation for oxygen gas: $2\text{KClO}_3(\text{s}) \rightarrow 2\text{KCl}(\text{s}) + 3\text{O}_2(\text{g})$ CHAPTER 9 REVIEW ...

CHAPTER 9 REVIEW - Doral Academy Preparatory School

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Reaction stoichiometry uses molar relationships to determine
the amounts of unknown reactants or products from the
amounts of known reactants or products. CHAPTER 9 DO NOT
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Chapter 9: Stoichiometry (Regents can skip all of section 9.3)

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•Lab 8: Quantitative Analysis •Regents Tables : Table T :
Important Formulas and Equations •Warm-ups and problems will
be collected before you take the test. Answer all problems in the
space provided.